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ADVANCED FUEL QUALITY ASSURANCE STANDARDS BASED ON THERMAL TESTING & CHEMOMETRIC MODELING

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IASH 14th Int'l. Symposium Charleston, SC Oct 2015

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- Steve Westbrook, George Wilson (SwRI)
- Joel Moreno, Indresh Mathur (Haltermann Solutions)



Outline



- Motivation/Background
- Approach
- Referee Fuel Set
- Thermal Integrity Test
 Method and Results
- Model Development and Results
- Future Work





Motivation: Fuel Quality Assurance



- Propulsion fuel performance, quality, and suitability must be verified
- This challenge is faced by:
 - Aerospace propulsion development/demonstration activities
 - Agencies who procure fuels for DoD use
 - Fuel manufacturers and suppliers
- Many requirements to consider:
 - Propellant cost
 - Support operations/infrastructure
 - Product availability & sustainability
 - Functional performance: combustion, cooling, lubrication...

Fuel thermal stability and material compatibility

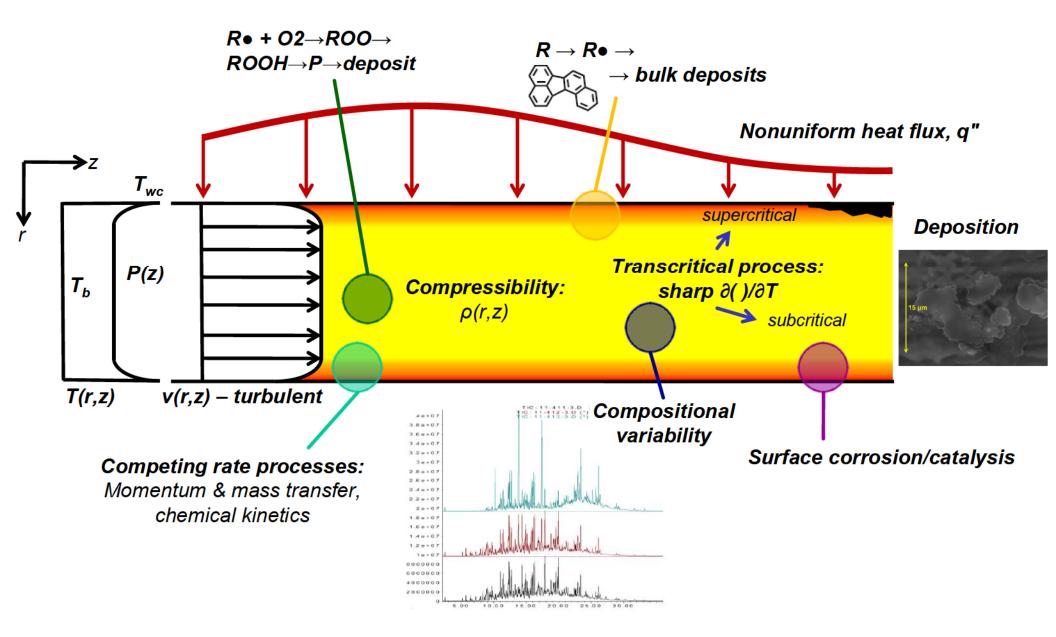
Aerospace Cooling System Conditions and Environments

Application	T _{wall} (°F)	T _{fuel, bulk} (°F)	Pressure (psi)	Heat Flux (Btu/in²s)	Material
Rockets	500-900	100-500	700-7000	10-120	Cu alloys
Hypersonics	1200-1500	100-1300	500-1000	0.5-2	Ni alloys
Aircraft	300-400	100-300	500-800	<1	SS alloys



Liquid Rocket Engine (LOX/Kerosene) Regenerative Cooling Environment







Background 1. Fuel Specification



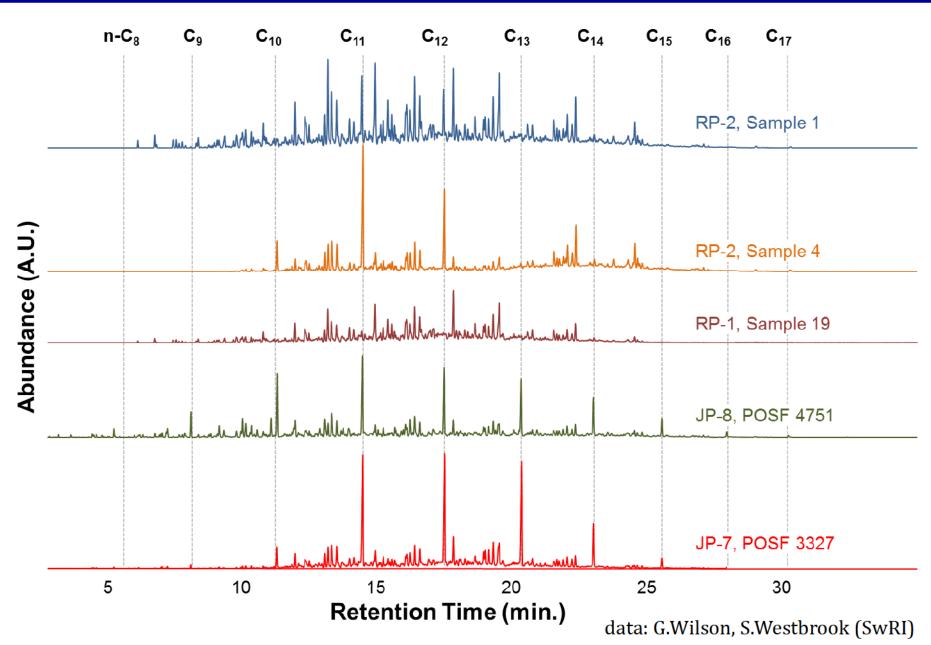
	ASTM Method	JP-5	Jet A	RP-1	RP-2	
Charification		MIL-DTL-	ASTM	MIL-DTL-	MIL-DTL-	
Specification		5624U	D1655-15	25576E	25576E	
Requirement, Units						
Distillation, °C						
IBP		report		report	report	
10% recovered		<205	<205	(185-210)	(185-210)	
20% recovered	D86	report				
50% recovered	D00	report	report	report	report	
90% recovered		report	report	report	report	
End point		<300	<300	(<274)	(<274)	
Doneity/15°C kg/I	D1298	0.788-	0.775-	0.799-	0.799-	
Density/15°C, kg/L	D1296	0.845	0.840	0.815	0.815	
Viscosity/-20°C, mm ² /s	D445	<8.5	<8.0	<16.5 ^b	<16.5 ^b	
Flash Point, °C	D93 ^c	>60	>38	(>60)	(>60)	
Freezing Point, °C	D2386d	<-46	<-40e	(<-51)	(<-51)	
Net Heat of Combustion,	varies ^f	>42.6	>42.8	(>43.0)	(>43.0)	
MJ/kg		10.1	4 0. 4 h	,	,	
Hydrogen, mass %	varies	>13.4	>13.4h	>13.8	>13.8	
Aromatics, vol %	D1319	<25.0	<25.0	<5	<5	
Olefins, vol %	D1319			<2.0	<1.0	
Total sulfur, mass%	varies ⁱ	<0.3	<0.3	<0.003	<0.00001	
Mercaptan sulfur, mass%	D3227	<0.002 ^j	<0.003 ^j	< 0.0003		
Thermal Stability: ΔP change, mmHg	D3241 ^k	<25	<25		report	

- Specification review and development activities are important for fuel qualification
- Physical, chemical spec limits are influenced by operational factors:
 - Performance
 - Handling/storage
 - Cost/Availability
 - Neither engine performance *nor* fuel chemical composition are specified *per se...*



Background 2. Compositional Variation







Background 3. (Lack of) Thermal Performance Test



ASTM D3241 (JFTOT) Results

(Shaded fuels shown in previous slide)

325°C, 5 hr. ←	→ 355°C, 5 hr.
	1

Fuel Type	JP-7	JP-8	RP-1	RP-2	RP-2	RP-TS-5	RP-2	Fuel Type	RP-1	RP-1
Designation	POSF	POSF	Sample	Sample	Sample	Sample	Sample	Designation	Sample	Sample
Designation	3327	4751	19	4	1	14	6		18	X
Tube Deposit	<2	>4APa	-2	-2	-2	-2	-2	Max ΔTDR,	35-38	۲
Rating Code	~ <u>~</u>	>4AP"	<2	<2	<2	<2	<2	spun	$(17)^{b}$	3
Maximum	0.1	200.1	0.1	0	0.1	0	0	Maximum	0	0
ΔP, mmHg	0.1	280.1	0.1	U	0.1	U	U	ΔP, mmHg	U	U

^a "A" denotes abnormal deposit; "P" denotes peacock deposit.

b Filtered

data: G.Wilson, S.Westbrook (SwRI)

R.Cook (AFRL), M.Thiede (AFPET)

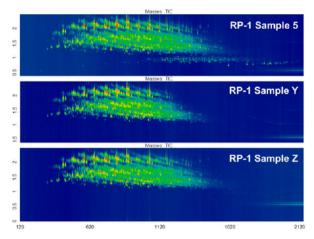
- Rocket kerosene is not quality tested for thermal stability or material compatibility prior to delivery
 - RP-1 is not tested. RP-2 is tested with ASTM D3241 (JFTOT) but results are "report only"
- JFTOT method may be valuable for screening very low performing fuels (contamination, alternative sources)...
- But the method is inadequate for ensuring fuel quality as increasingly demanding thermal environments arise



AFQTMoDev Project Structure

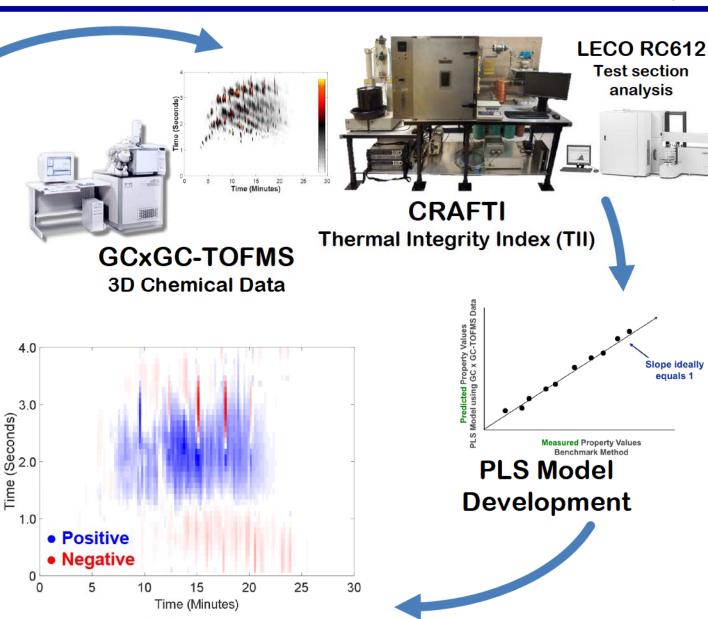


Fuel Compositional Variation



1. Optimize Composition

2. Specification Limits



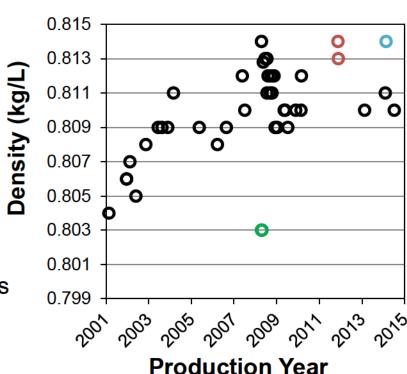
Compound Correlation to TII



Referee Fuel Set



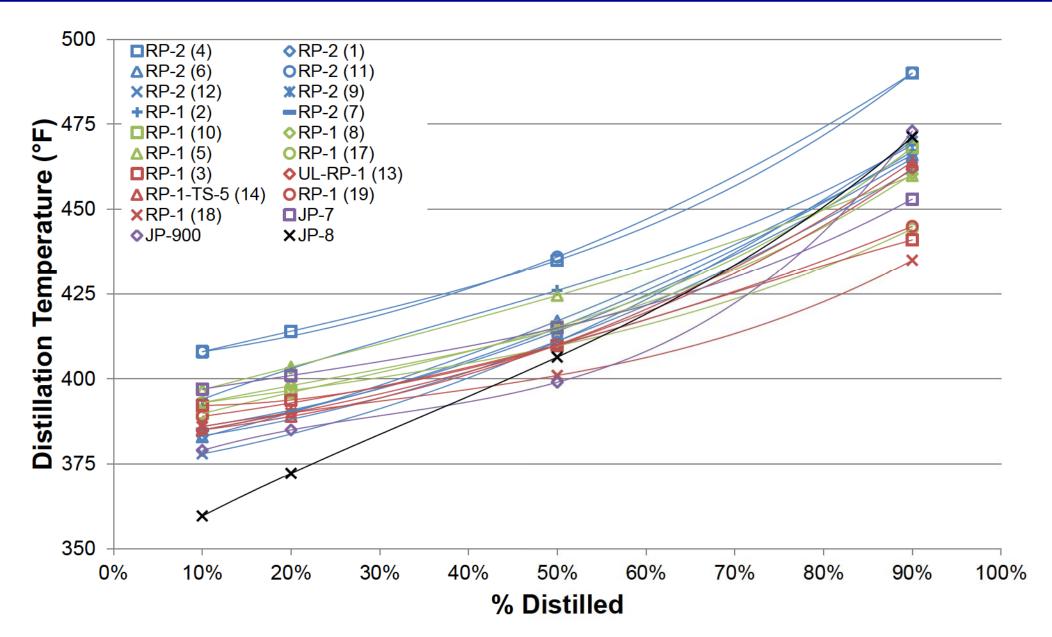
- Criteria for fuel selection
 - Multicomponent: distribution of hydrocarbon species and/or types
 - Possess heteroatom species diversity
 - Span the compositional range of fuels meeting MIL-DTL-25576E: not necessarily "today's fuel"
 - Meet aerospace fuel designations for health/flammability/reactivity, etc.
- What we ended up with
 - 51 compositionally unique fuels (or potential blend materials – single/multicomponent)...
 - 19 evaluated for thermal integrity and included in chemometrics/modeling
 - 。 (8) RP-2,(9) RP-1, JP-7, JP-900
 - 3 available from previous SwRI project
 - Less than ideal compositional variation
 - Produced on demand no repository of historical fuels
 - Relatively consistent production past 20 years
 - Several "interesting" fuels contained common feedstocks





Referee Fuel Set Variation

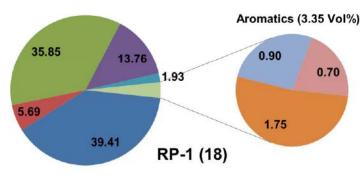


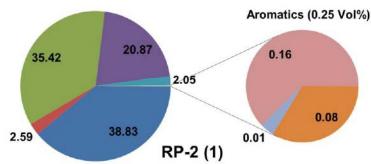


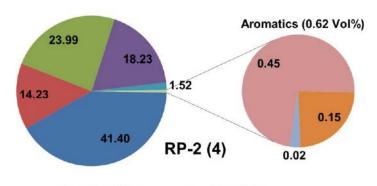


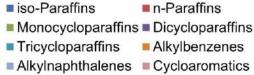
Fuel Set Compositional Variation

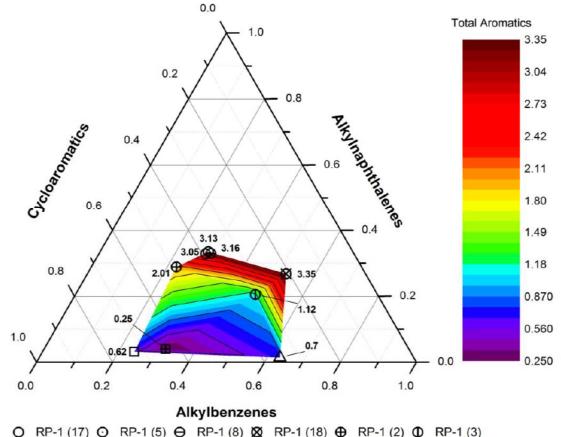










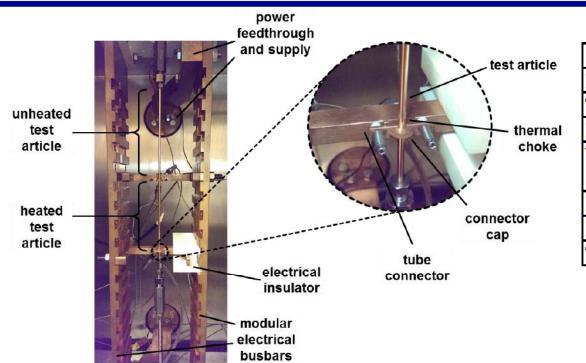


O RP-1 (17) ⊙ RP-1 (5) ⊖ RP-1 (8) Ø RP-1 (18) ⊕ RP-1 (2) Φ RP-1 (3) ⊞ RP-2 (1) □ RP-2 (4) △ JP-7 (15)



Compact Rapid Assessment of Fuel Thermal Integrity (CRAFTI)

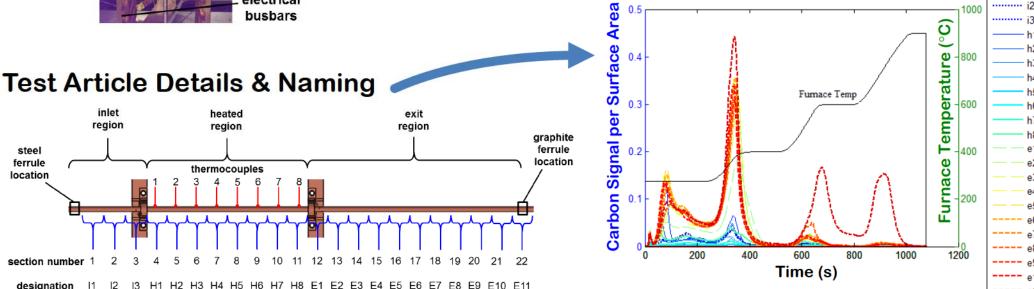




Standard Test Conditions

Parameter	Value	Units
Reynolds Number, Re	2000-20,000	-
Test article material	Cu (C10100)	-
Input power	4500	W
Wall temperature	~1050±250	°F
(dependent variable)	(560±120)	(°C)
Backpressure	1,000 (6.9)	psi (MPa)
Heated length	4 (10.2)	in. (cm)
Test duration	15	min.

Test Article Analysis Temperature Programmed Oxidation





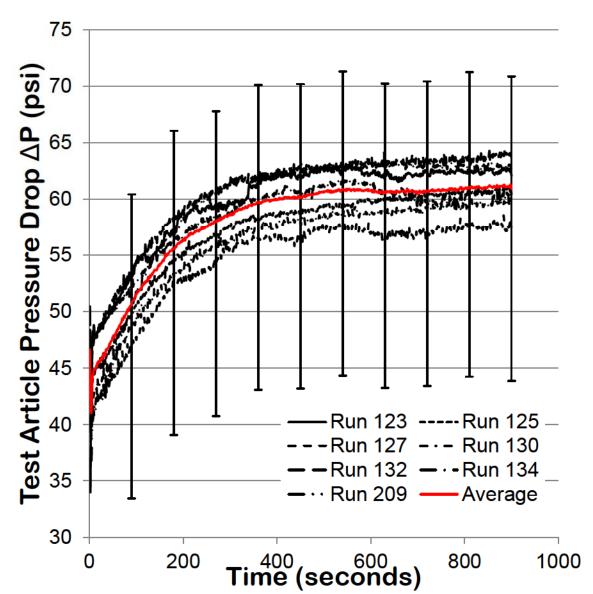
Repeatability: Pressure Drop Increase



Ten runs were performed at standard test conditions using baseline fuel (RP-2 Sample 1)

- 6 runs initially
- 2 runs 2/3 mo. later
- 2 runs 9/10 mo. later
- Purge/flush/purge protocol between fuels; no disassembly
- Pressure drop can be indicative of deposit formation
 - Variation from other sources should be minimized
- Pressure drop variability from test to test was well within measurement uncertainty $\delta(\Delta P) \cong \delta(P_{in}) + \delta(P_{out})$

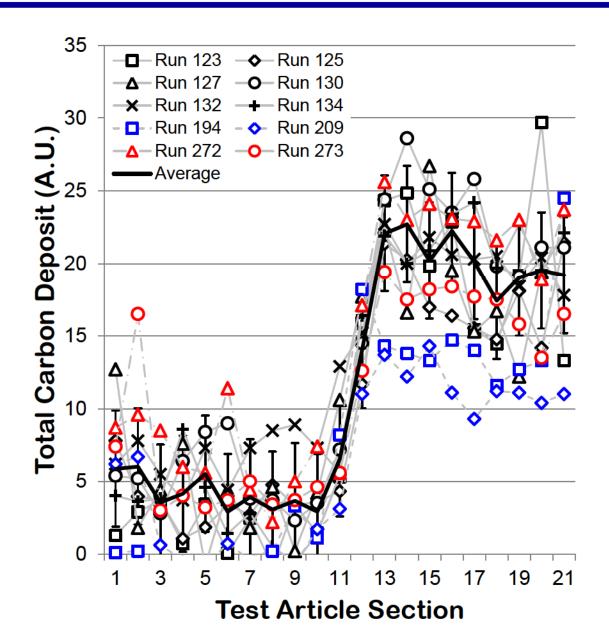
RP-2 Sample 1 (7 Runs Shown)





Repeatability: Deposit Formation





- For ten runs with baseline fuel, carbon deposit behavior is similar – and initially somewhat unexpected
- Near detection limits → some noise likely due to instrument response
- These results indicate "end-to-end" variation (fuel, experiment, test article handling, analysis, etc.)
- Will carbon deposit behavior vary with fuel composition?



CRAFTI Results Indicate Method Sensitivity for Thermally Stable Fuels



CRAFTI v1.1 Conditions/Results (15 min., T_{o,bulk} ~ 650°F, T_{wc} ~ 800-1200°F)

(Shaded fuels: Indistinguishable ΔP with JFTOT)

Fuel	Sample #	# of Runs	Average Wall Temperature °F (°C)	Pressure Drop ΔP, initial psi (kPa)	ΔP Increase during Test psi (kPa)
RP-2	1	10	1158 (626)	44 (306)	16 (113)
RP-2	4	4	1026 (552)	42 (288)	20 (136)
RP-2	7	2	1048 (564)	39 (269)	21 (142)
RP-1	3	7	1112 (600)	46 (320)	21 (146)
UL-RP-1	13	2	1115 (602)	33 (225)	22 (154)
RP-2	9	4	1110 (599)	31 (213)	23 (158)
RP-2	6	2	1145 (618)	47 (326)	23 (158)
RP-1	10	2	1057 (569)	31 (217)	26 (179)
RP-2	12	2	1144 (618)	29 (203)	30 (207)
RP-1	19	2	1074 (579)	26 (180)	30 (210)
RP-2	11	2	1085 (585)	30 (205)	33 (225)
RP-TS-5	14	2	1130 (610)	32 (222)	35 (242)
JP-900	16	2	1015 (546)	31 (217)	45 (313)
JP-7	15	2	950 (510)	30 (205)	50 (343)
RP-1	8	2	964 (518)	35 (241)	79 (545)
RP-1	5	2	985 (529)	41 (281)	82 (563)
RP-1	2	7	1027 (553)	45 (311)	91 (625)
RP-1	17	1	964 (518)	31 (217)	103 (713)
RP-1	18	2	1005 (541)	30 (210)	188 (1293)

- Standard test conditions produce measureable performance differences
- Pressure drop increase varies from 40-630% of initial value
- Most fuels meet current RP-1/RP-2 limits (MIL-DTL-25576E)
- JFTOT results indicated indistinguishable performance (ΔP increase after 5 hours) for these fuels

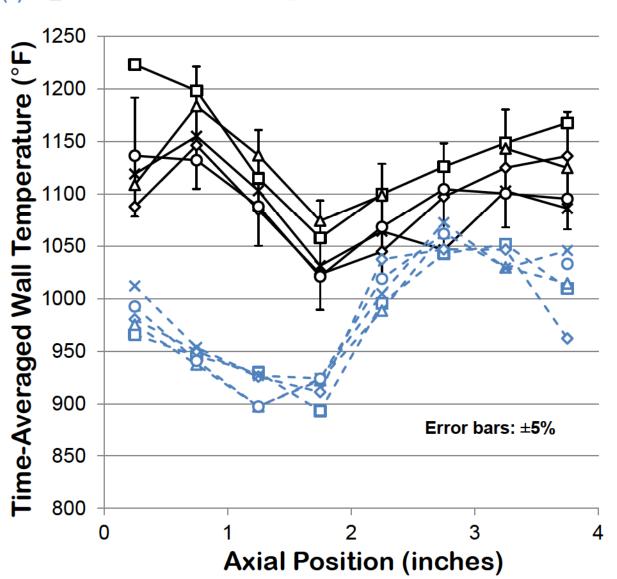


Wall Temperature Behavior (Heated Region Only)



RP-2 (1): —□—Run 123 —◆—Run 125 —▲—Run 127 ——Run 130 —○—Run 134 RP-1 (2): —□—Run 152 —◆—Run 154 —▲—Run 158 ——Run 160 —○—Run 164

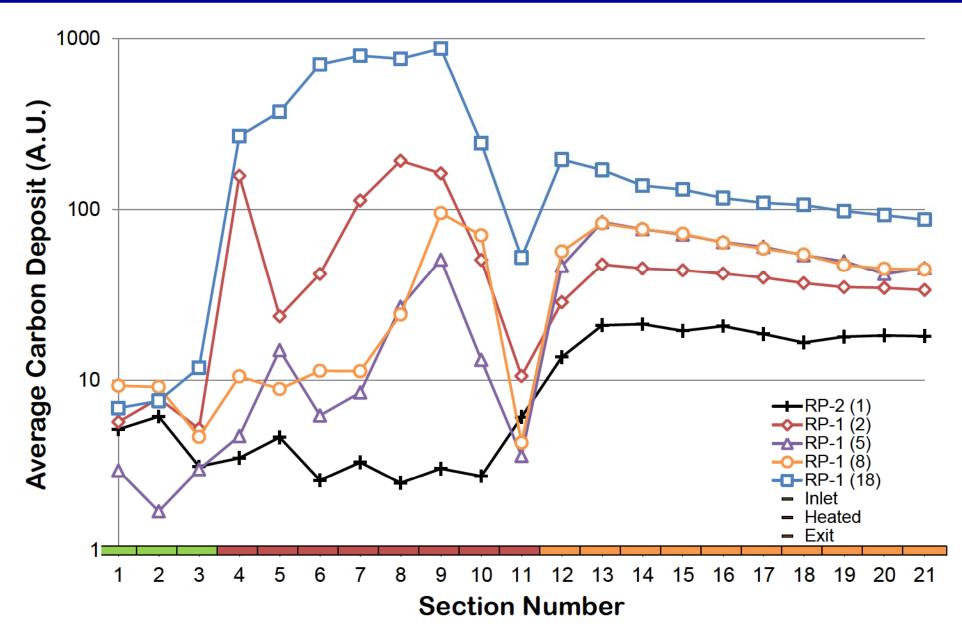
- Wall temperature can be indicative of fuel thermal integrity, but is complicated by other factors:
 - Electrical connection → local current flux density
 - Deposit formation → effects on local heat transfer
 - Transcritical flow → property gradients
- Repeatable characteristic profile for fuels of different thermal quality...
- Difficult to explain temperature/ time history variation
- Modeling & simulation underway to characterize fluid/solid thermal environment, flow behavior





Time-Integrated (Total) Carbon Sensitive to Fuel Composition

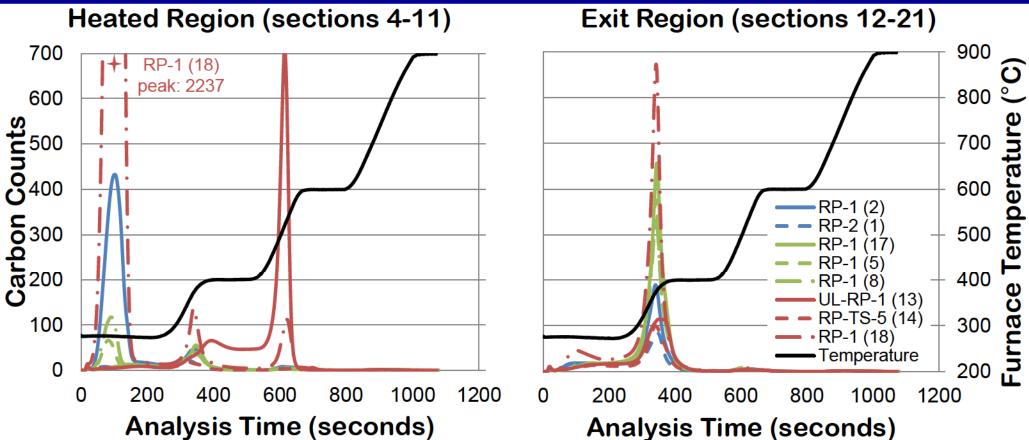






Differentiated Carbon Data Provides Additional Insight





- Highest depositing fuels showed largest pressure drop increase
 Exception: UL-RP-1 (13): significant deposit but small ΔP increase
- Heated region carbon deposits predominantly chemisorbed (0-200s)
- Amorphous carbon (200-450s) dominates exit region
- Only one fuel with strong filamentous carbon signal (450-800s): UL-RP-1 (13) in heated region
- Pressure drop increase correlated with amorphous deposit in exit region?



LECO

Fuel

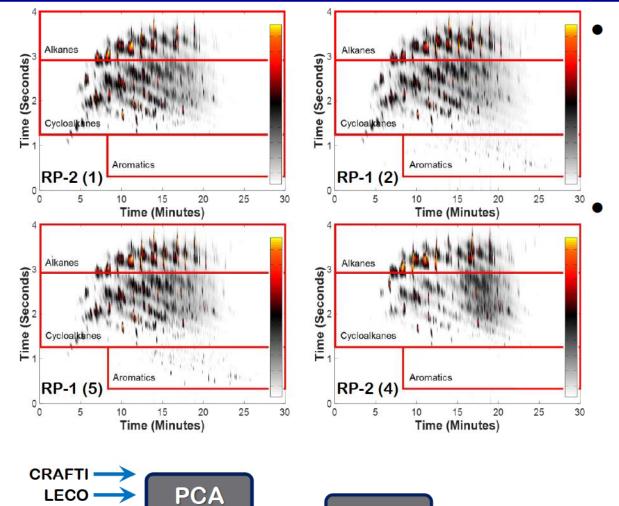
Compositional

Differences

GC×**GC**

Chemometrics with CRAFTI, Carbon Deposit, & Comprehensive GC×GC-TOFMS Datasets





F-ratio

Group

Distinguishing

Compounds

Purpose of chemometrics: Clarify role of fuel composition in cooling performance/quality
Guide fuel formulation

- Advise specification methods/limits

Implementation:

- Principal component analysis (PCA)
 - Assign categorical quality
 - Identify important compositional differences
- Fisher ratio (F-ratio) analysis
 - Refine GCXGC dataset for optimized models
 - Identify distinguishing chemical compounds
 - Partial least squares (PLS) modeling
 - Develop predictive models that relate thermal integrity behavior to fuel composition

PLS

Predictive Models



20

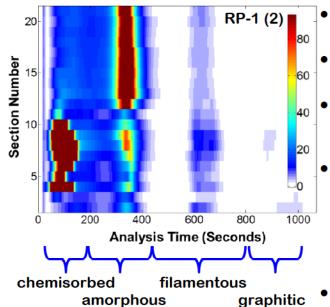
200

Analysis Time (Seconds)

Section Number

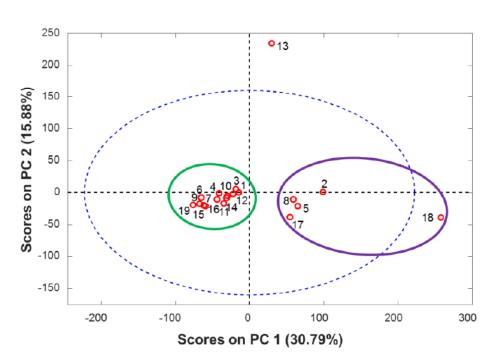
PCA Example: Correlate Carbon Deposit Types/Regions with Channel ΔP Increase





- Map of 3D TPO data
- Available for 19 referee fuels
- Multivariate data: excellent PCA candidate
- How can information be made useful?
- PCA PC1 Loadings Plot
- Associates positive (blue) & negative (red) contributions to PC1 with original data matrices
- Positive contributions to PC1 (blue) correlate with high pressure drop
- ΔP most sensitive to amorphous carbon (200-450s) in exit region (sections 12-21)

- PCA Scores Plot: PC groupings capture variance in measured data (ideally 100%)
- In this case, high △P fuels (purple) and low △P fuels (green) group together – primarily along PC1
- A relationship between carbon deposit and pressure drop is confirmed – but what does PC1 represent?



Similar analyses performed for test article pressure drop, wall temperature, GCXGC-TOFMS chromatographic variation

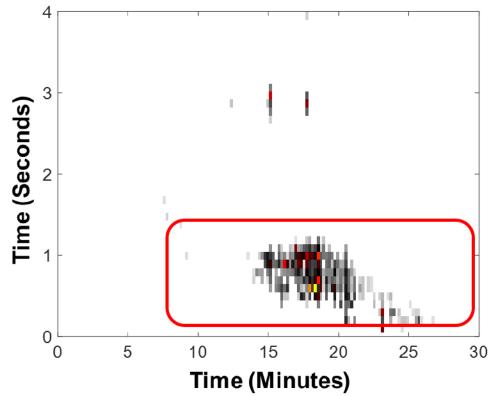


Fisher Ratio Analysis: Identify Compounds Responsible for Group Assignment



#	F-	t_r^1	t_r^2	Compound	Match	C-
L"	Ratio	(min.)	(sec)	Compound	Value	ratio
1	277.1	18.7	0.91	1,1,6-trimethyltetralin	908	13.0
2	254.1	18.3	0.97	5-ethyltetralin	846	15.9
3	233.0	18.7	1.01	(1,4-dimethylpent-2- enyl)benzene	769	16.9
4	231.4	18.8	0.96	1-methyltetralin	751	6.3
5	219.4	16.5	1.20	(1-ethylbutyl) benzene	790	14.5
6	214.0	15.6	1.31	5-methylnonane	840	1.7
7	201.3	18.1	1.31	1,3,5-trimethyl-2- propylbenzene	894	11.9
8	188.5	23.6	0.64	2,6-dimethyl naphthalene	935	68.3
9	188.2	18.2	3.31	2,6-dimethyl heptadecane	900	5.0
10	182.9	17.3	1.43	Adamantane	885	2.4
11	179.4	17.6	1.29	1-ethyl-2,4,5- trimethylbenzene	792	16.0
12	177.6	23.5	0.41	Biphenyl	926	34.8
13	175.9	18.5	1.31	6-propyltetralin	735	9.1
14	172.4	18.9	0.81	6-methyltetralin	955	16.4
15	171.4	19.1	0.91	2,3-dimethyltetralin	711	4.6
16	169.0	17.5	1.31	1,4-dimethyl-2-(2- methylpropyl)-benzene	825	13.3
17	167.9	18.8	1.13	1-heptenylbenzene	794	10.1
18	164.6	17.7	1.19	(5-methyl-1- hexenyl)benzene	764	13.8
19	162.7	18.8	1.33	(1-methylhexyl) benzene	818	18.4
20	161.9	15.4	1.25	p-cymene	814	6.9

Top 300 F-Ratio Chromatographic Locations



- F-Ratio Analysis top hits are primarily aromatic...
- But their relative influence is quantified
- Hits represent class-distinguishing compounds, not necessarily direct influences on fuel thermal integrity
- Reduces superfluous chemical data in PLS model development



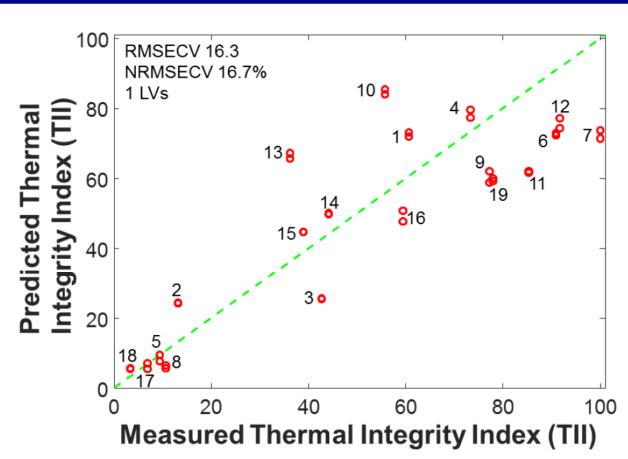
PLS Modeling: Predict Thermal Integrity using GC×GC-TOFMS Data



- Leave One Out Cross Validation (LOOCV):
 - N models generated, each with N-1 datasets
 - Datasets include CRAFTI, TPO, and GC×GC data
 - Each resulting model used to predict behavior for fuel left out during model generation
 - Statistical/graphical comparison of predicted vs. measured values
- Top 300 F-Ratio tiles used
- New parameter defined based on PCA, F-Ratio efforts: Thermal Integrity Index (TII):

∘
$$TII \propto (\Delta P_{\text{max}} \times C_{\text{A,exit}})^{-1}$$

- Good model agreement
 - Not inclusive of all compositional influences
 - Does not account for ΔP_{init}



Similar predictive models developed for test article pressure drop, wall temperature, carbon deposit



Summary



- A compositionally diverse set of rocket kerosene fuels was acquired and systematically evaluated
- A compact, rapid fuel thermal integrity assessment (CRAFTI apparatus) was developed and used to quantify fuel performance. Qualification criteria:
 - Operates at conditions relevant to intended application
 - Produces meaningful data quickly using small fuel quantity
 - Performance data collected with adequate repeatability
 - Discriminate between otherwise indistinguishable fuels
 - Results are traceable to existing experiments
 - Possesses characteristics of a standard test method
- Chemometric analyses applied to multiparametric datasets
 - Improvements in understanding of physicochemical influences and impacts of deposit formation were made
 - Predictive, composition-based models were developed these models are adaptable to additional datasets and expandable to diversified fuel sets:
 - Pressure drop, wall temperature, carbon deposit, etc.

Thank You for Your Attention







